

Preservation Research and Testing Division (PRTD) Research



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Library of Congress

Development of New Preservation Technologies

(increasing access to collections)

- Requirement for Library collection materials to be on (long-term) or semi-permanent exhibit
- **Inter-Agency Agreement** with the National Institute for Standard and Technology (NIST)
- Abel Buell Map of America (1784)
- Collaboration with NIST and NARA - develop oxygen sensor

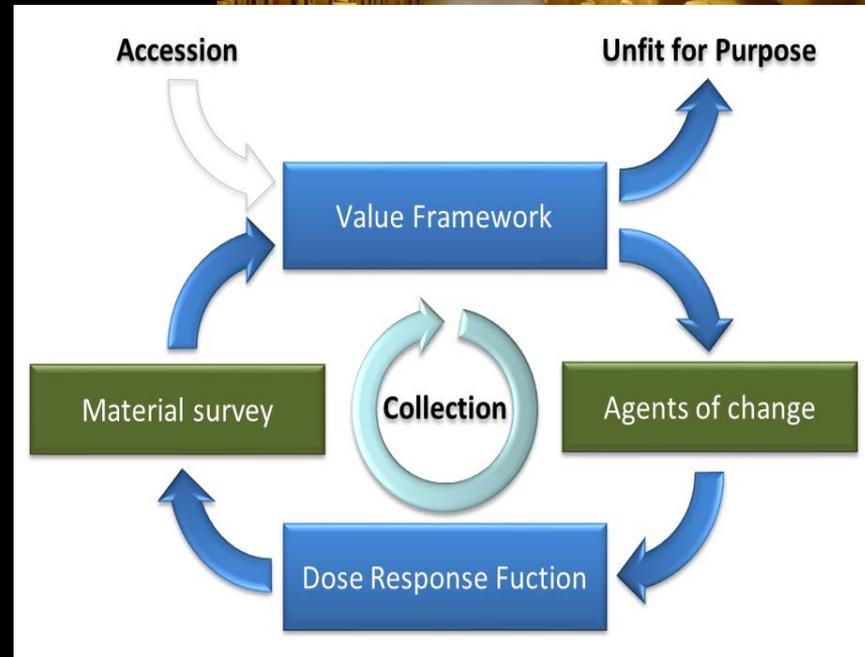




Collections Demography

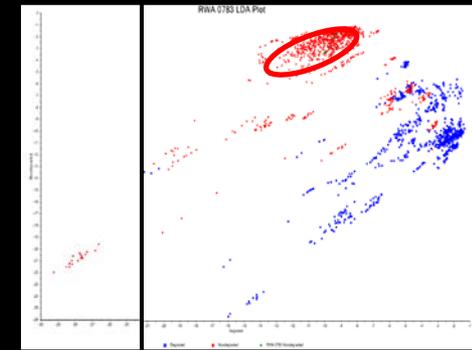
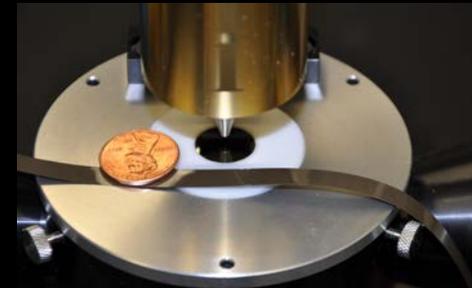
(collection preservation and management)

- Collaboration with University College London, Centre for Sustainable Heritage
- *Demography* –analysis of the size, distribution, and characteristics of a population
- Assessing specific materials as “collection” and user response
- Linking usage of collections with physical change
 - Handling reduced molecular weight of cellulose polymer
 - Developing non-invasive method to assess changes in collection materials



Degradation of Magnetic Tape

- LC as non-funded partner on Institute for Museum and Library Services (IMLS) 3year grant (ending 2016)
- To develop a tool for libraries, archives and museums to detect (before damage) whether a tape can safely be played for transfer and access to content
- Rapid non-destructive identification of degraded magnetic tape
- Final component to expand to wider range of tape formulations



Classification as non-degraded or degraded



Magnetic Tape Degradation

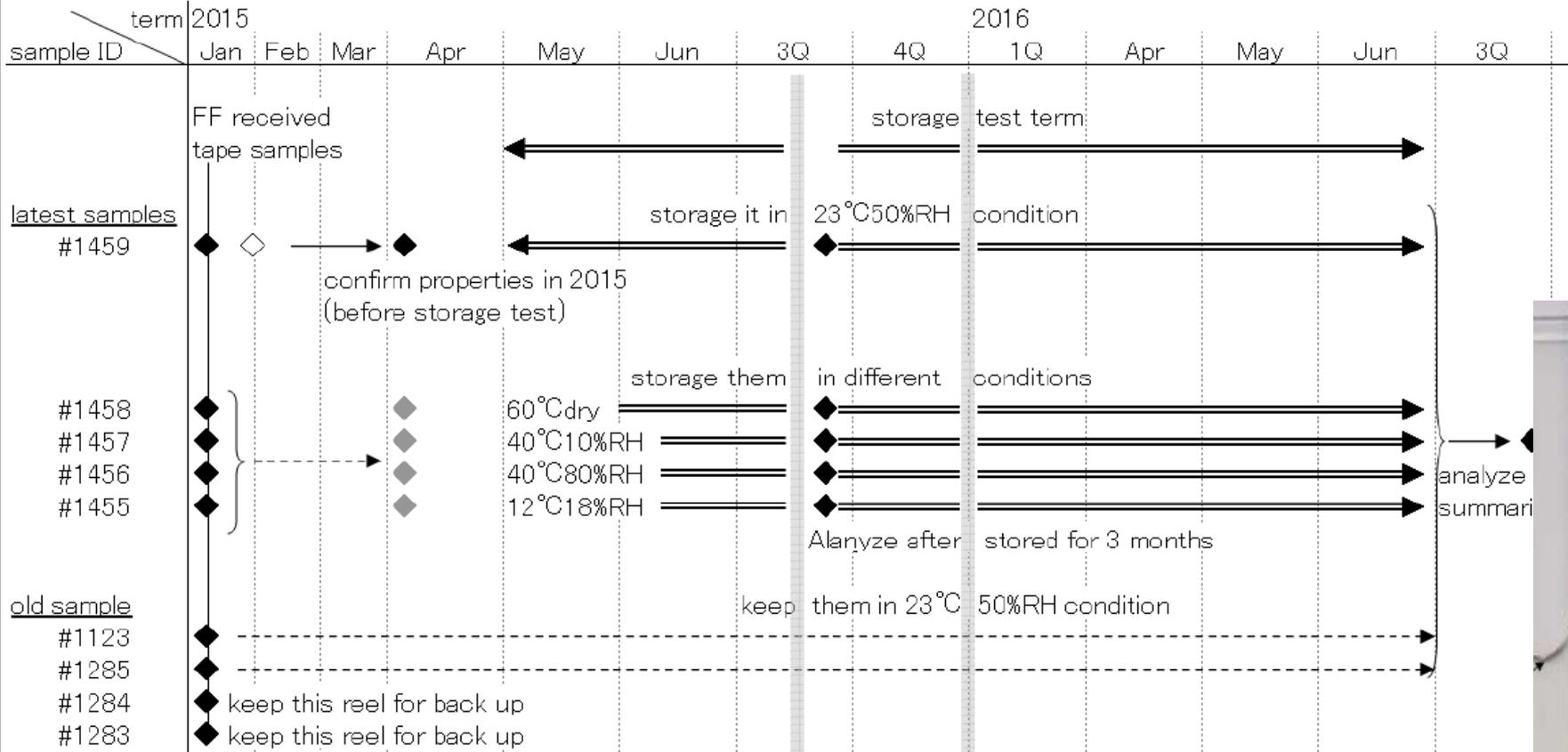
- Memorandum of Understanding (MOU) between LC and FujiFilm to utilize their accelerated aging laboratories to induce degradation of older tape formats
- Ongoing aging to assess chemical and physical properties in range of storage environments
- Tracking change over time



FUJIFILM

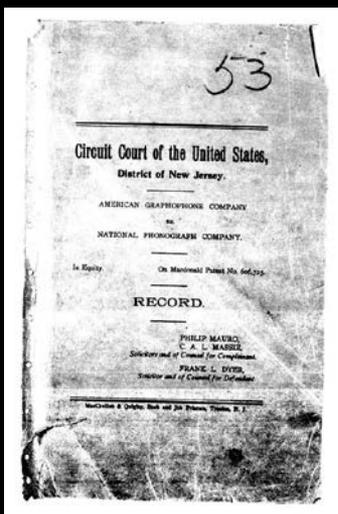
Storage test plan-1

- The latest proposal of the schedule

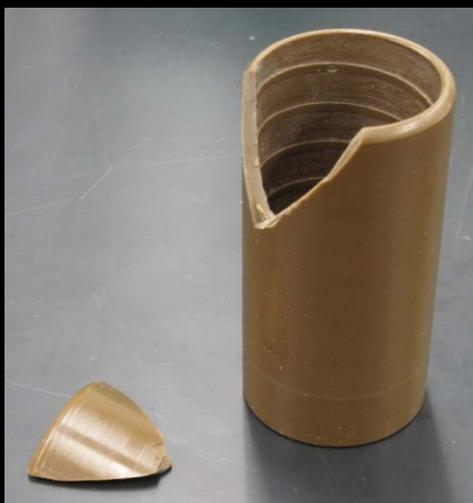


Because Fujifilm will make shutdowns in summer and winter holidays, it will take a little longer than actual 12 month. Fujifilm would like to propose adding the analysis of 3 months storage in August to the analysis of 1 year storage.

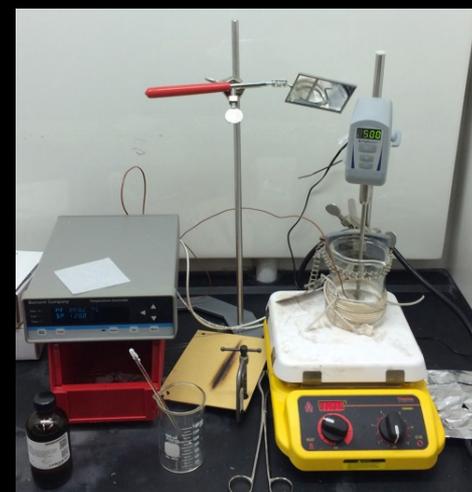
Deterioration of Wax Cylinders



Historical Documents



Chemical and Physical Analyses



Laboratory Synthesis

Understanding condition, composition, degradation and storage options

Lawsuit re: MacDonald Patent 606,725

Legal Box 170

IN THE
Circuit Court of the United States

DISTRICT OF NEW JERSEY.

AMERICAN GRAPHOPHONE COMPANY,
Complainant.

vs.

NATIONAL PHONOGRAPH COMPANY,
Defendant.

} In Equity.

Brief for Defendant.

FRANK L. DYER,
MELVILLE CHURCH,
for Defendant.

PAKES OF W. F. GORRETS COMPANY, WASHINGTON, D. C.

1905-1907

14

COMPLAINANT'S PROOFS.

Comparing the process of the patent in suit and that described in the patent No. 782,375, and paralleling the formulas given we have the following:

Macdonald Patent in Suit.

- (1) stearic acid, 300 pounds;
- (2) caustic soda, 9 pounds;
- (3) sal soda, 60 pounds;
- (4) aluminium, 1.5 pounds;
- (5) ceresin.

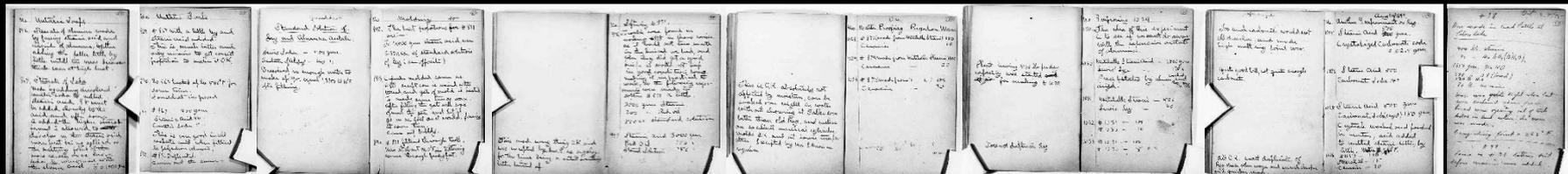
Aylsworth Patent No. 782,375.

- (1) stearic acid, 100 pounds;
- (2) caustic soda, 452 gr. 1 lb.
- (3) sal soda (sodium carbonate) 21.5 pounds;
- (4) aluminium, 175 gr. equals
*.385 pounds;
- (5) ceresin;
- (6) carnauba wax.

10. Defendant further answering, on information and belief, says that the said Letters Patent herein sued upon are invalid and void, because the same were obtained surreptitiously and unjustly for an invention which was, in fact, invented by another or by others who were using reasonable diligence in adapting and perfecting the same, namely:—

By Thomas A. Edison at West Orange, New Jersey, now residing at Llewellyn Park, Orange, New Jersey.

By Jonas W. Aylsworth at West Orange, New Jersey, now residing at East Orange, New Jersey.



Experiment 1058

Aug 1889 #87

Too much carbonate would not all dissolve, and made high melting point wax.

Quite good but, not quite enough carbonate.

all O.K. exact duplicate of Reg made other ways and much cheaper and quicker made.

Aug 14/1889, (1058)

No. Another Improvement on Reg.

1055. Stearic Acid 500 gm.
Crystallized Carbonate soda 2.025 gm.

1056 Stearic Acid 500
Carbonate Soda 140

1057 Stearic Acid 500 gm
Carbonate Soda (crist) 150 gm.
Crystals washed, and powdered in mortar, and added to melted stearic little, by little, Melted at 230° F.

1058 #1057 - 100
Stearic Acid - 15
Cerasin - 30

Experiment 38 (NB 1000)

Oct 1895 #38 Oct 2.4/95

Wax made in lead bottle at Silver Lake (77)

400 lb. stearic
92 " $\text{Na}_2\text{CO}_3 (10\text{H}_2\text{O})$
1359 gm. Na_2O
598 " Al (powd)
120 lb. H_2O
30 lb. cerasine.

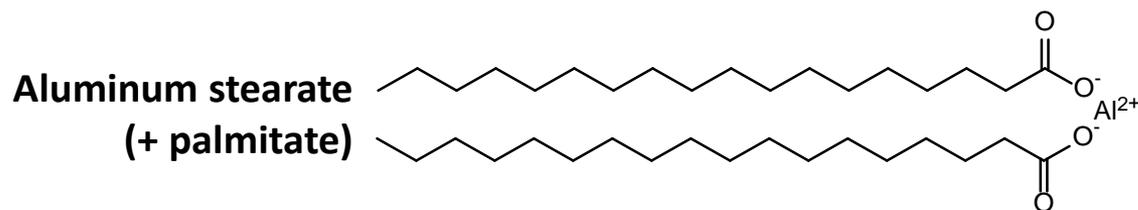
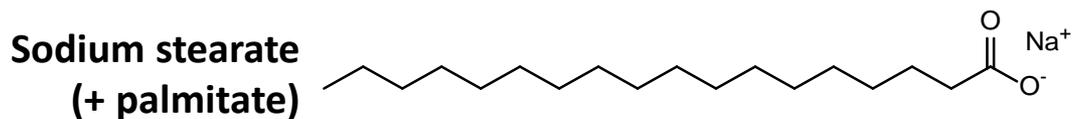
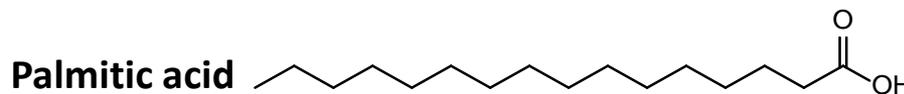
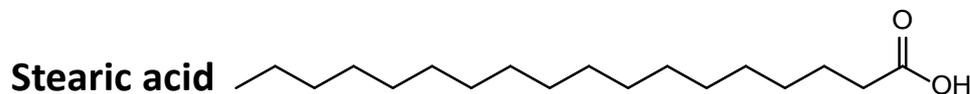
Wax was quite light color but was darkened some from burnt wax oozing out of tooth holes in lead when the wax was made.

Coagulating point = 262° F

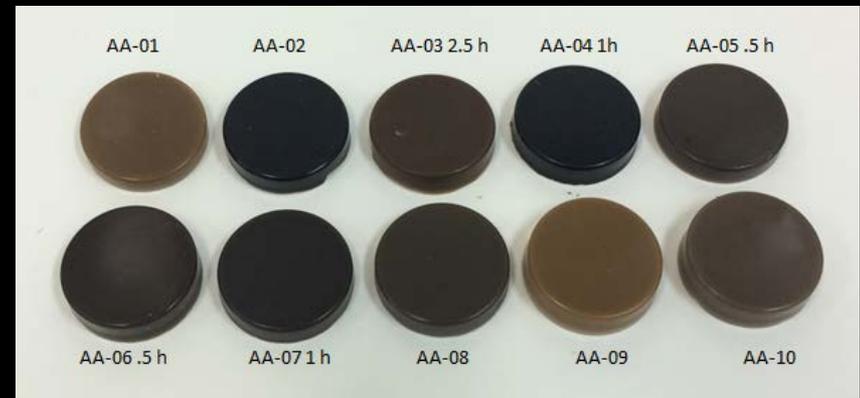
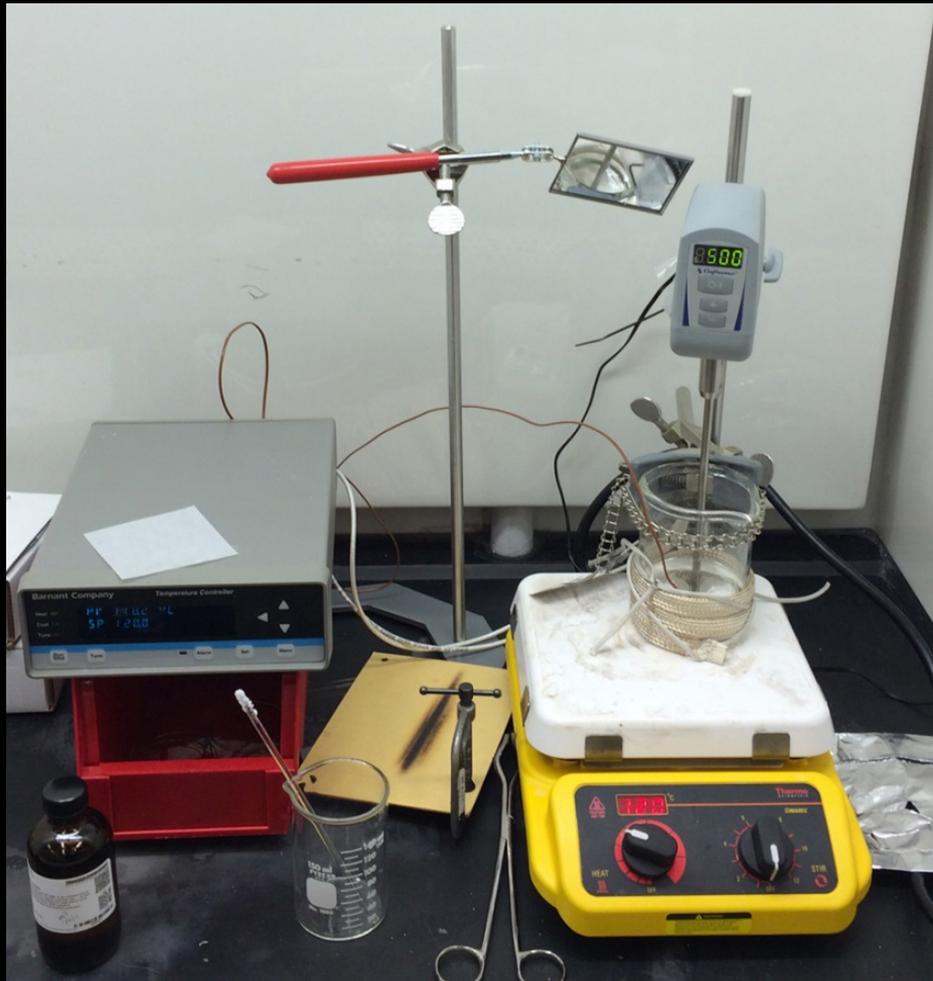
#39

Same as #38 taken out before cerasine was added.

Chemical Composition from Lab Notebooks

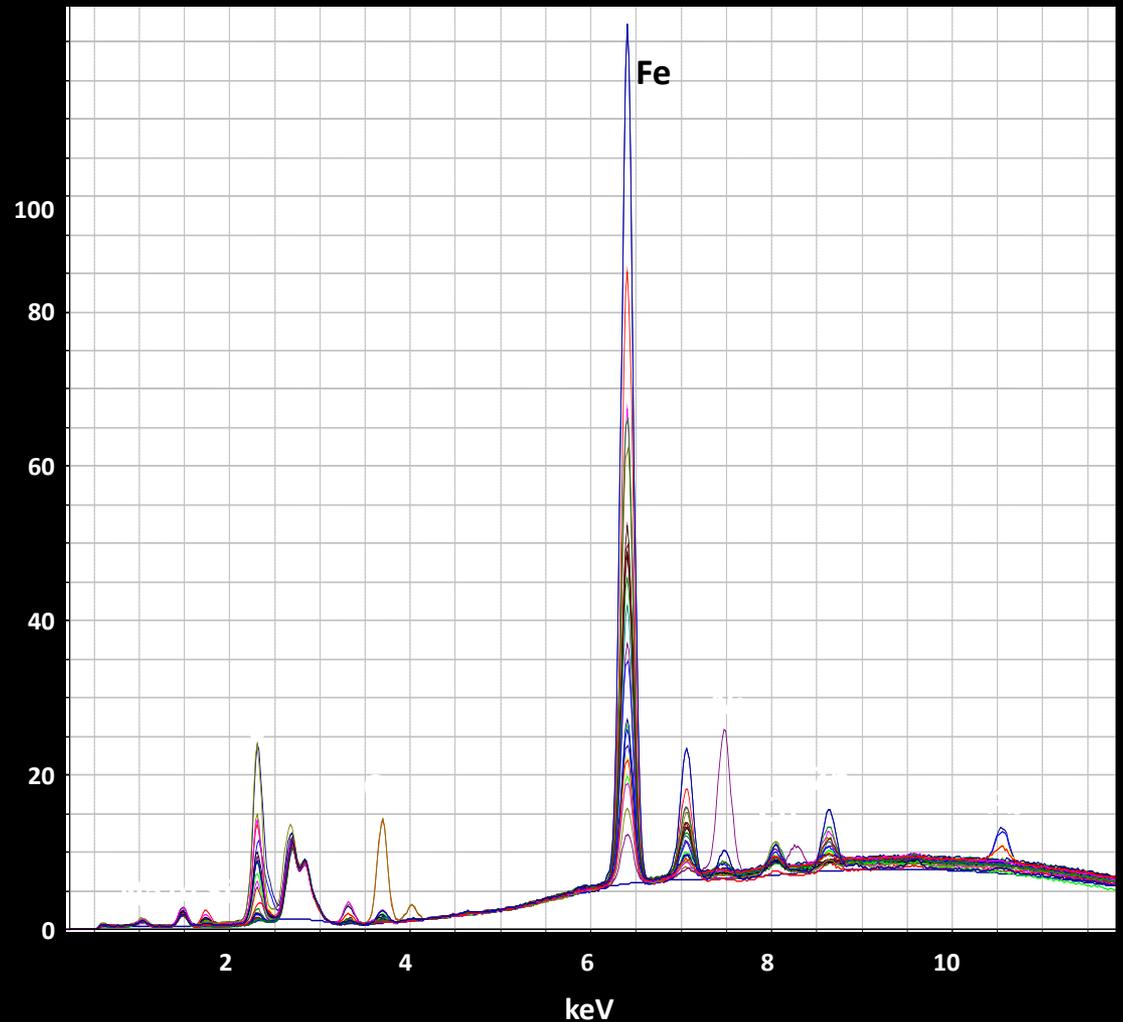


Producing Waxes from Historical Recipes

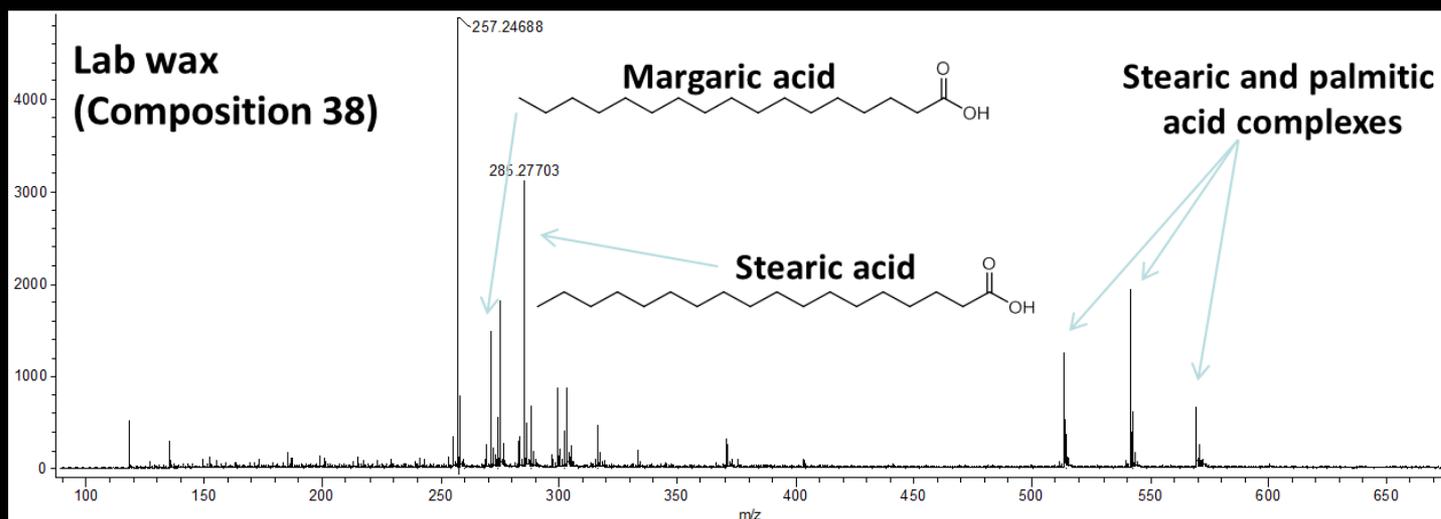
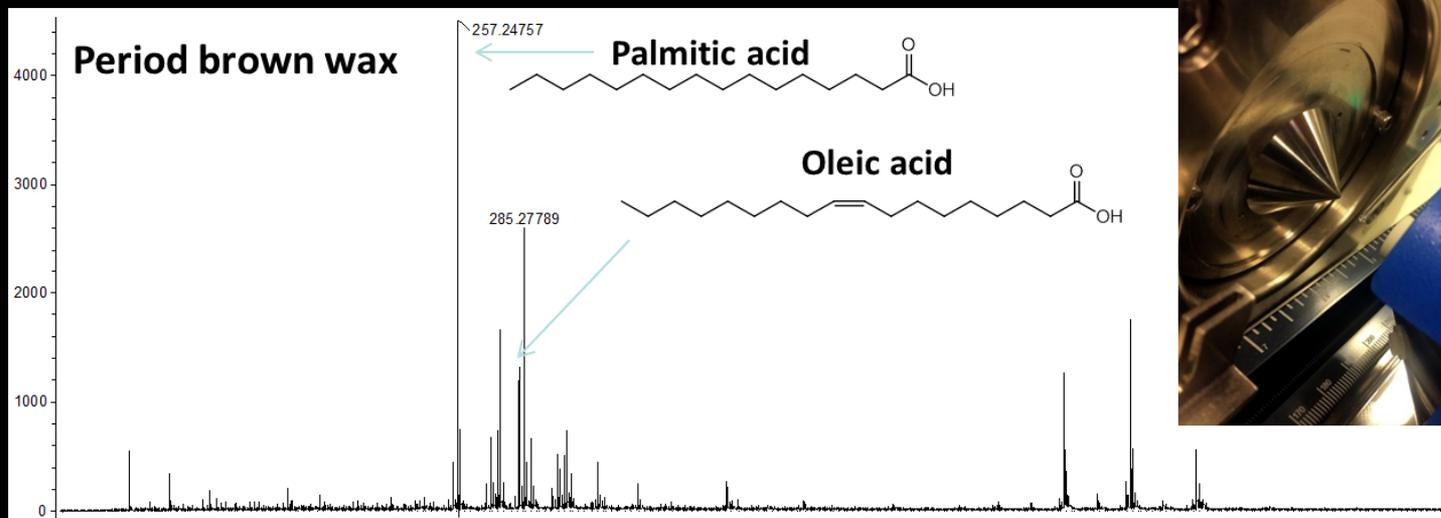


Non-invasive Analysis of Original Cylinders

Inorganic (trace metals)



Analysis of Organic Composition



Thermal Expansion: Impact on Cylinders

